Viscosimetric Molecular Weights: Some Computer Programs for Experimental Data Elaboration

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Synopsis

New methods have been developed for the computer which allow viscosimetric data elaboration. Two methods evaluate the intrinsic viscosity, the Huggins or the Kraemer constant, the molecular weight, and their confidence limits according the Student's t-distribution starting from the simplest viscosimetric data. The kinetic energy influence is considered. Another method is used when the Huggins constant is known.

INTRODUCTION

The viscosity of dilute solutions is very often used for measuring polymer molecular weights because the experiments are simple and reliable. The graphic elaboration of the experimental data according to Huggins¹ or Kraemer² for obtaining the intrinsic viscosity, from which through the Mark-Houwink relationships^{3,4} molecular weights are evaluated, is quite tedious and subject to errors. Computer programs have already been prepared for accelerating and improving the treatment of data,^{5–8} however, these programs made use of simplified formulas and partially elaborated data.

In this paper, we present programs which evaluate the intrinsic viscosity, the Huggins (or Kraemer) constant, the molecular weight, and their confidence limits according to the Student's *t*-distribution starting from the simplest viscosimetric data such as the polymer weight, the solvent volume, and the solvent and solution efflux times.

GENERAL RELATIONSHIPS

The specific viscosity η_{sp} and the relative viscosity η_{rel} are so defined⁹⁻¹¹ considering the kinetic energy influence^{12,13}:

$$\eta_{sp} = \eta_{rel} - 1 = \frac{\rho t}{\rho_s t_s} \left[\frac{1 - \frac{K}{t^2}}{1 - \frac{K}{t^2_s}} \right] - 1$$
(1)

where ρ_s and ρ are the solvent and solution densities, t_s and t are the efflux 2583

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1+	DIMENSION (1000)**(1000)**(1000)*IITLE(13)*ANI(10)*VOLSOL(1000)*#
2+	1ESQL(1000) +R0SQLU(1000) +CC(1000)
3+	63 READ(5+80)MN
4 +	BO FORMAT(I3)
5+	D0163 II=1+MN
6+	READ(5+60) (TITLE(J)+J=1+13)
7*	WRITE(6,59)(JITLE(J)+J=1+13)
8.	60 FORMAT(13A6)
10.4	27 FURMAI(1010/04)1040 PEAN(5-2)/ANT(4).4-1.4).70500
11+	2 FORMAT(6F10, 3, 110)
12+	READ (5+61) AC+AKA
13+	61 FORMAT(F19, 4+F10, 3)
14+	62 READ (5+B)NN+AKK+ROSOLY-ROPOLY
15+	8 FORMAT(13+3F15-8)
16*	IF(NN.LE.1)6070163
17+	READ(5+9)TS+(T(I)+I=1+NN)
18*	9 FORMAT(10F8-3)
19*	READ(5,99) WEIGHT, (CC(I),I=L+NN)
20*	YY FORMATCH 10. /+10F/. 3/
228	
23*	SUNCC=SUNCC+CC(N)
24+	VOLSOL(M)=SUNCC
25+	X(M)=WEIGHI+100.0/SUMCC
26*	77 CONTINUE
27+	SUNX=0.0
28*	SUNY=0.0
29*	SUM XY =0.0
30*	
32+	
33#	
34+	WESOL (N) = VOL SOL (N) + ROSOL V
35+	ROSOLU(M)=(WEIGHT+WESOL(M))/(VOLPOL+VOLSOL(M))
36+	Y[#]=R050LU(#]+T(M]+(1.0-AKK/T(M]++2)/((RD50LV+TS+X(M))+(1.0-AKK/T
37+	1S++2))-1+0/X(H)
38+	444 IF(AKK+LT=0+0)Y(M)=(T(M)-TS)/(TS+X(M))
39+	SUNX=SUNX+X (N)
40.*	
41*	
434	30mag-30mag/a(m/*A(m) 7 continue
	HRI IE(6, 19)
45+	19 FORMAT(///.1X+* LIST OF MEASURES *./)
46+	IF(AKK .L T. 0. 0) 60 T0666
47+	WRITE(6+667)TS
48+	667 FORMAT(//+2X+*TS =*+F7.2+5X+* WITH DENSITY AND KINETIC-ENERGY CORR
49+	IECTIONS (+//)
50*	IF(AKK.GI.0.0)6910777
524	DDD WRIIELDFDDD/IS AAR Eddwarf/2,14,4 Is -9,57 3,54,4 uituaut carrentars 4,771
538	111 NURCHN 000 LORMANIA, 12 - 46195424, WILHOUL CORRECTIONS (111)
54+	C=NN
55+	XMED=SUMX/C
56+	N I = NN - 2
57*	B=(SUMXY-XHED+SUMY)/(SUMXQ-C+XHED++2)
58+	A=SUNY/C-B+XMED
59*	PK=8/A**2
6U#	
614	NA30 M−1-NAGC
674	
64+	QMIN=QHIN+(Y1-Y(M))++2
65*	XQ=XQ+{X(M)-XMED)++2
66*	WRITE(6+20)M+T(M)+X(M)+Y(M)+Y1
67+	20 FORMAT(1X+12+2X+* TIME(SEC)=*+F7.2+2X+* X(G/DL)=*+F7.4+2X+* Y=*+F7
68+	1.4+2X+* Y1=*+F7.4+/+)
69*	30 CONTINUE
70+	STR=SQRT(QHIN/(C-2+0))
71*	AL LABE ANI (NI) F () TX/ SUM () XU/) Al Masani (NI) F () TX/ SUM () XU/)
/2*	AL 1MA=AN 1/N 1/F 37 A FOUNT 1 (0/ 0/ANEUF 2/ AV/
734	21 FORMAT(//+ STANDARD DEVIATION=*+F7.4)
75+	WRITE(6, 39) IPERC
76+	39 FORMAT(/// CONFIDENCE INTERVAL =*+13)
77+	WRITE(6+5)A+ALIMA
78+	5 FORMAT(//.* INTERCEPT(=INTRINSIC VISCOSITY)=*+F6.3+3H +-+F6.4+//)
79+	WRITE(6+69)B+ALIMB

Fig. 1 (continued)

80+	69 FORMAT(1X+* SLOPE=*+E10+4+3H +-+E10+4+/+)
81*	BMAX=B+ALIMB
82*	BMIN=B-ALIMB
83+	AFAX=A+ALIMA
84+	AMIN=A-ALINA
85*	PKMAX=BMAX/AMIN++2
86*	PKHIN=BMIN/AMAX++2
87+	WRITE(6+233)PK+PKMAX+PKMIN
68*	233 FORMAT(/+ HUGGINS CONSTANT=++F7.4+2X+ MAX=++F7.4+2X+ MIN=++F7.
89+	14+)
90+	IF(AKA.LT.0.0)601073
91+	AGEA/AKA
92*	RA=1.0/AC
93*	AL = ALOG(AO)
94+	ADA={A+AL [HA]/AKA
95+	AOR= (A-ALIMA)/AKA
96*	ALA=ALOG(AOA)
97*	AL H=ALDG(ADH)
98+	ARGERAAAL
99*	AMEEXP(ARG)
100+	AMA=EXP(RA+ALA)
101+	ANI=EXP(RA+ALN)
102*	WRITE(6+79)AC+AKA
103*	79 FORMAT(//+1X+* A=*+E10.4+4X+* K=*+E10.4+///)
104+	WRITE(6+6)AN+ANA+ANI
105+	6 FORMATCIX+* AVERAGE #01 WEIGHT=*+E10.4+2X+* MAX=*+E10.4+2X+* MIN=*
1064	
10/#	
100+	
110*	FND
	Fig. 1. First computer program using Huggins equation.
-	
1 *	DIMENSION TELOOODAACIOUTTELOODATTELECTISTAATELOOTTELSEETE
2*	
3*	63 READ(5+80) HN
4+	BO FORMAT(13)
5.	D0165 11=1+MN
6*	
7*	WEIE(6,54)([]][C(3)+3-14(3)
6*	60 FORMAT(1)46) FO FORMAT(1)41-39-1144)
¥#	77 FURNALLANI (2011)AU (
10-	S CADMATIAFIA JA JA
117	
124	KEAU(7+6]) AC 4AKA
134	61 FURHATEFIN. 49E 10.53
	62 READ(5+63)NN+AKK+K3SSEV+KOPOLT
17*	8 TUMPAI(15)5 T0.8)
10+	
1/#	REAU(3+9) 5+((1)+1=1+NN)
184	9 FURMAI(10FB.3)
144	READ(5)99)WEIGHIJ(CC(1))I=L,NN)
204	94 FURNATION 10. 1. 10F / . 4)
22#	
23#	
244	
25#	
26#	
27*	SUM X=0-0
28.	SUNY=0-0
29+	SUMXY=0+0
30+	SUMX9=0.0
31+	DO 7 M=1+NN
32*	IF (AKK.LT.0.0)6010444
33+	VOLPOL=WEIGHT/ROPOLY
34+	WESOL(M)=VOLSOL(M) +ROSOLV
35*	RDSDLU(W)=(WEIGHT+WESDL(M))/(VOLPOL+VOLSOL(M))
36*	Y(H)=ALDG(RDSOLU(H)+T(H)+(1.0-AKK/T(H)++2)/(ROSOLV+TS+(1.0-AKK/TS+
3/=	177717784777 686 76787777
70-	TTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTT
40+	
41+	
42*	
43+	7 CONTINUE
\$4+	WRITE(6+19)

Fig. 2 (continued)

45+	19 FORMAT(///+1X++ LIST OF MEASURES ++/)
46*	IF (AKK.LT.0.0)6070666
47=	WRITE(6+667)TS
48+	667 FORMAT(//+2X+*TS =*+F7+2+5X+* WITH DENSITY AND KINETIC-ENERGY CORR
49+	1ECTIONS ++//)
50*	IF(AKK.6T.0.0)60T0777
51*	666 WRITE(6+668)TS
52*	66B FORMAT(//+1X+* TS =*+F7+2+5X+* WITHOUT CORRECTIONS *+//)
53*	777 NOBSENN
54+	C=NN
224	
574	₩1 = ₩₩ ₩2 = ₩₩5 ₩ 5 / / / / / / / / / / / / / / / /
584	D-1 50741747820*3077771507489-648788*727 A-5188776-8549850
59.	
60+	X9=0-0
61+	
62+	0330 M=1+N285
63+	Y1=A+B+X(M)
64+	QMIN=QMIN+(Y1-Y(M))++2
65*	XQ=XQ+(X(M)-XNED)++2
66*	WRITE(6+20)M+T(M)+X(M)+Y(M)+Y1
67*	20 FORMAT(1X+I2+2X++ TIME(SEC)=++F7.2+2X++ X(G/DL)=++F7.4+2X++ Y=++F7
68*	1_4+2X+* Y1=*+F7_4+/+)
69*	30 CONTINUE
70+	SYX=SQRT(QMIN/(C-2.9))
71+	ALIMB=ANI(NI)+(SYX/SQRT(XQ))
72*	ALIMA=ANI(NI)+SYX+SQRT(1.0/C+XMEQ++2/XQ)
73+	WRITE(6+21)5YX
74+	21 FORMAT(//+ STANDARD DEVIATION=+F7.4)
75*	WRITE(6.39)IPERC
76*	39 FORMAT(///* CONFIDENCE INTERVAL =*+13)
774	WKI ELEFEJA FALIMA E pomatyje u tuteosotjetutotuste viscositvi-tela 2020 an.e.e. 4.//i
70+	DELETION INTERCEPTICIALINICALE ALTONIAL ALTONAULA - ALGOARTA
80.	WT1161010770785100 40 FORMAT174.0 51 OD5-1.510.4.34 4.510.8.7.)
81.	
828	
83+	AMAX=A+ALTNA
84+	AMINEA-ALIMA
85+	QKMAX=BMAX/AMIN++2
86*	QKMIN-BMIN/AMAX*+2
87+	WRITE16+233)QK+QKMAX+QKMIN
88+	233 FORMAT(//+* KRAEMER CONSTANT=*+F7+4+3X+* MAX=*+F7+4+3X+* MIN=*+F7
89*	1.4)
90+	IF (AKA.LI.0.0)601073
91*	A0=A/AKA
92*	
934	AL = AL DGTADJ
944	AUA-IATALINA/IATA
934	
97.	
98+	ARGERATAL
99.	AM=EXP (ARG)
100+	AMA=EXP(RA=ALA)
101+	AHI=EXP(RA+ALM)
102*	WRITE(6+79)AC+AKA
103*	79 FORMAT(//+1X+* A=*+E10_4+4X+* K=*+E10_4+///)
10**	WPITE(6+6)AM+AMA+AMI
105+	6 FORMAT(1X+* AVERAGE MOL WEIGHT=**E10+4+2X+* MAX=*+E10+4+2X+* MIN=*
106+	1+E10.4+/1HO)
107*	73 6010 62
108*	163 CONTINUE
109*	
110.	ENU

Fig. 2. Second computer program using Kraemer equation.

times of the solvent and of the solutions from the capillary viscometer, K is the viscometer constant, which depends on its geometry.

The concentration of diluted solutions is generally very low (<1 g/100 cc) so that the solvent and the solution densities are very close. Supposing that $\rho_s = \rho$, if K/t_s^2 and K/t^2 are both negligible with respect to unity, which can be obtained with suitable viscometers, eq. (1) becomes

$$\eta'_{sp} = \eta'_{rel} - 1 = \frac{t}{t_s} - 1.$$
 (2)

The equations which correlate η_{sp} and η_{rel} to the concentration c are

$$\eta_{sp} / c = [\eta] + k' c [\eta]^2$$
(3)

$$\ln \eta'_{rel} / c = [\eta] + k'' c [\eta]^2$$
(4)

where k' is the Huggins constant, k'' is the Kraemer constart, 2 and $[\eta]$ is the intrinsic viscosity.

1+	С	INTERPOLATION ATHAISQUARE #X
5+		DIMENSION T(1000)+C(10)3)+X(1000)+Y(1000)+Y(1000)+DIF(1000)+ XK(3)
3+		1,5Q(3),TITLE(13),R(4),XI(13),CC(100)
	89	READ(5+90)MN
5.4	91	FORMAT(T3)
		D0163TE1+MN
74		PERD(5,3) = (TTTEE(J) + J=1+13)
	1	
	-	
9*	-	
1)*		
11+	63	2 READ (5+6) NN+PK+AC+AKA
12•		IF(NN.LE.1)3070163
13+		FORMAT(13+2F10+4+E10+3)
14 •		READ(5+6)T5+(T(M)+M=1+VN)
15+		6 FORMAT(10F8.3)
16+		READ(5+61)WEIGHT+(CC(M)+M=1+NN)
17 *	61	1 FORNAT(F10.7+10F7.4)
10-		
144		
20*	_	SUNCC=SUNCC+CC(N)
21 * 9	•	C(M)=WEIGHT+100+0/SUMC;
22*		X(P)=PK+C(P)
23•		Y(M)=(T(M)-TS)/(TS+C(M))
24+	7	7 CONTINUE
25•		SUMX=0.0
26*		SUMXY=0.0
27 •		SUNX0=0.0
28*		SUMY=0.0
29.		00 51 J=1+NN
3.0.		
31 .		
3.24		
334		
344		
37 .		B(2)=(1.5+SUAX)/SUAXQ
56*		QN=NN
37 •		B(3)=(QN-2+(0+SUMXY)/(2,0+SUMXQ)
38*		B(4)=-SUMY/(2.0+SUMXQ)
39 *		CALL SEQ3(B+XR+XI+KU+L)
# 0 *		DO 10 I=1+3
41 # 👘		IF(XI(I)-LT-0-0)60T010
42*		IF(XI(I).6T.0.0)60T010
43 *		SQ(I)=0.0
44+		BO 11 J=1+NN
45.0	1	1 SQ(1)=SQ(1)+(Y(J)-YQ(1)-YQ(1)+YR(1)+Y(J))+(Y(J)-YR(1)-YR(1)
	-	
		TE (1, FO, 2) AMTH-SA(T)
		1 L+ L+ L+ C / H = H = J + I + J + I + J + J + J + J + J + J +
50*	1	
21 4		OMIN=SQ(1)
52*		RMIN=XR(1)
53¢		00 30 K=2+3
54*		IF (SQ (K).GT.QMIN) GOTO3(
55 e		QMIN=SQ(K)
56+		RMIN=XR(K)
57 +	30	D CONTINUE
58+	44	O GRMIN=RMIN+RMIN
59 e		AD=RMIN/AKA
60+		AL=AL06(A0)
61 •		RA=1.0/AC
62+		ARGERA+AL
		NORTHOLDS.

Fig. 3 (continued)

63 e 6 4 #		PM=EXP WRITE((ARG) 6+80)				
65 *	80	FORMAT	{///•1ו	LIST OF MEAS	URES +/)		
66+		WRITE(6+81)75+	PK			
67 🔹	81	FORMAT	(//+1X+ *	1S=*#F7.2+5X+	 HUGGINS 	CONSTANT=",F	10+4,//,}
68*		D035M=	1 + NN				
69 *		¥1(M)=	RMIN+GRM	IN=*(4)			
70+		DIF(F)	=Y(M)-Y1	(円) (四):()(1)、(1)(1)、		1.115(#)	
<i>n</i> •		WRITE	0+022841	1	113) 1110		
72*	35	CONTIN	10E 144.19.17			24.1 016/011=	1. F7. A. 2 X. 1 X(=P
734		- FURMES		· · · · · · · · · · · · · · · · · · ·			
		IK WU J-		** T=**F7*4*2X = · · _ · ·	•• •I=••F	/+4+2X+* UIF-	***/***
754		IF (L.E	6.1)AKI1	L(6+84)			
77.	8 4	FURRAI	(//,1X,**	NUMBER J- REA	L X0312=3	*••	
774	704	IF (L+E	Q.21WK11	t(6+321)		. .	
734	321	PURMAI	(//+1/+*	NUMBLE J- HEA	L K0312-1	•• /	
80+	01	WRIIL(0.0314411	**************************************		NOTO NTOCOST T	V1-1.57
814		1-1-57	A - 27 - 7 W			1310 11500311	TF="#F F#### 2 ^# * #
82 *		SOTOAS			HOL WEIGH		
87.	143	CONTIN					
84 +	99	STOP					
85*		END					
L	POLYIS	SOBUT W	LENE A	IN CYCLOHEX	ANE T=3	OC VISCOM.	L
T	5 = 323.	.50	WITH	DENSITY AND	KINETIC	-ENERGY CORI	RECTIONS
1	TIME	SEC)=	731.10	X(G/DL)=	•3125	Y= 4.7933	¥1= 4.8084
2	TIME	SEC)=	575.50	X(G/DL)=	-2083	Y= 4.5025	¥1= 4.4916
3	TIRE	SEC)=	505.50	X(G/DL)=	.1563	Y= 4.3718	¥1= 4.3332
٩	TIME	(SEC)=	437.20	X (G/DL) =	.1042	Y= 4.1403	¥1= 4.1748

STANDARD DEVIATION= .0389

CONFIDENCE INTERVAL = 95

INTERCEPT(=INTRINSIC VISCOSITY)= 3.858 +- .2282

SLOPE = .3041+01 +- .1087+01

HUGGINS CONSTANT: .2043 MAX: .3133 MIN: .1171

A= .6900+00 K= .2760-03

AVERAGE MOL WEIGHT = . 1018+07 MAX = .1107+07 MIN = .9322+96 Fig. 4. OUTPUT of the first program according to cards 3-8 of the INPUT of Table II. POLYISOBUTYLENE & IN CYCLOHEXANE T=30C VISCOM.1

LIST OF MEASURES WITHOUT CORRECTIONS TS = 323.50 TIME(SEC) = 731.10 Y1= 4.0468 1 X(G/DL)= .3125 Y= 4.0319 TIME(SEC) = 575.50 X(G/DL)= .2083 Y= 3.7391 Y1= 3.7265 2 3 TIME(SEC)= 505.50 X(G/DL)= •1563 Y= 3.6006 Y1= 3.5663 TIME(SEC)= 437.20 X(G/DL)= .1042 Y= 3.3741 ¥1= 3.4061 STANDARD DEVIATION= .0360 CONFIDENCE INTERVAL = 95 INTERCEPT(=INTRINSIC VISCOSITY)= 3.086 +- .2109 SLOPE = . 3075+01 +- .1004+01 HUGGINS CONSTANT= .3229 HAX= .4935 MIN= .1905 A= .6900+00 K= .2760-03 AVERAGE HOL WEIGHT= .7368+06 MAX= .8108+06 MIN= .6649+06 Fig. 5. OUTPUT of the first program according to cards 10-15 of the INPUT of Table II.

Experimentally the viscosimetric measurements are repeated varying the initial concentration with dilution: plotting eqs. (3) and (4), one obtains $[\eta]$ which is correlated to polymer molecular weight M through the Mark-Houwink^{3,4} equation

$$[\eta] = KM \tag{5}$$

where K and a are characteristic constants for every polymer-solvent system.

PRINCIPLES OF CALCULATION

First of all, the values of the different concentrations (in g/100 cc) are calculated from the polymer weight and from the solvent volume (cc) added in every test. Considering the additivity of the volumes, the solution densities are calculated from the polymer and solvent densities. According to eq. (1), η_{sp} is calculated from the values of K, t_s , and t for every concentration c. However, it is possible to use the simplified eq. (2), too. At this stage, all the

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   POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2
LIST OF MEASURES
                 WITH DENSITY AND KINETIC-ENERGY CORRECTIONS
TS = 204.30
1
    TIME(SEC) = 563.50
                        X(6/0L)= .3212
                                          Y= 6.2817
                                                      ¥1= 6.3016
2
    TIME(SEC) = 424.60
                        X(G/DL)= .2141
                                          Y= 5.8420
                                                      ¥1= 5.8150
    TIME(SEC) = 361.40
3
                        X(G/DL)= .1606
                                          Y= 5.5975
                                                      Y1= 5.5717
    TIME(SEC) = 302.40
                        X(G/DL)=
                                  .1071
                                          Y= 5.2957
                                                      ¥1= 5.3285
STANDARD DEVIATION= .0378
CONFIDENCE INTERVAL = 90
INTERCEPT(=INTRINSIC VISCOSITY)= 4.842 +- .1505
SLOPE= .4544+01 +- .6974+00
HUGGINS CONSTANT= .1938
                           HAX= .2382
                                         MIN= .1544
A= .6900+00
                 K= .2760-03
AVERAGE MOL WEIGHT= .1415+07
                                MAX= .1480+07
                                                 MIN= .1352+07
Fig. 6. OUTPUT of the first program according to cards 17-22 of the INPUT of Table II.
```

data to be entered into eq. (3) are available. Substituting now $y = \eta_{sp}/c$ and x = c, eq. (3) becomes

$$y = a_0 + a_1 x \tag{6}$$

where $a_0 = [\eta]$ and $a_1 = k'[\eta]^2$.

The best values of the constant a_0 and a_1 are obtained by a least-squares method. The confidence intervals (with the coefficient of confidence $1-\alpha$) for the intercept a_0 and for the slope a_1 are given¹⁴ from

$$a_{0} \pm t_{\alpha/2; n-2} s_{y/x} \sqrt{\frac{1}{n} + \frac{(\bar{x})^{2}}{\sum_{i=1}^{n} (x_{i} - \bar{x})^{2}}}$$
(7)

$$a_1 \pm t_{\alpha/2;n-2} \frac{s_{y/x}}{\sqrt{\sum_{i=1}^n (x_i - \bar{x})^2}}$$
 (8)

POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2

LIST OF MEASURES

```
TS = 204.30
                 WITHOUT CORRECTIONS
                                                        Y1= 5.4948
    TIME(SEC)= 563.50
                                           Y= 5.4738
1
                        X(6/DL)= .3212
    TIME(SEC)= 424.60
                        X(G/DL)= .2141
                                           Y= 5.0357
                                                        Y1= 5.0066
2
    TIME(SEC) = 361.40
                        X(G/DL)= .1606
                                           Y= 4.7881
                                                        ¥1= 4.7626
3
    TIME(SEC) = 302.40
                        X(G/DL) = .1071
                                           Y= 4.4848
                                                        Y1= 4.5185
STANDARD DEVIATION= .0392
CONFIDENCE INTERVAL = 90
INTERCEPT(=INTRINSIC VISCOSITY)= 4.030 +- .1559
SLOPE= . 4559+01 +- .7223+00
HUGGINS CONSTANT= .2807
                           MAX= .3518
                                          MIN= .2189
                 K= .2760-03
A= .6900+00
AVERAGE HOL WEIGHT= .1085+07
                                 MAX= .1146+07
                                                  #IN= .1025+07
Fig. 7. OUTPUT of the first program according to cards 24-29 of the INPUT of Table II.
```

where $t_{\alpha/2;n-2}$ is the Student's *t*-coefficient with n-2 degrees of freedom. The numerical values of the *t* coefficient are tabulated¹⁵ and reported in Table I as a function of the degrees of freedom n-2 (that is, also of the number of tests) and of α ; $1 - \alpha$ represents the probability that the values of a_0 and of a_1 are included between the limits indicated in eq. (7); \bar{x} is the average value of x_{ij} ; $s_{y/x}$ is the standard deviation

$$s_{y/x} = \sqrt{\frac{\sum_{i=1}^{n} (y_i - y_c)^2}{n-2}}$$
 (9)

where y_c is the calculated value of y.

Then, the calculations of $k' (= a_1/a_0^2)$, $M (= (a_0/K)^{1/a})$, and their maximum and minimum values in the considered confidence intervals are performed.

The FORTRAN IV program of this calculation is reported in Fig 1. Another program, which uses the Kraemer eq. (4), instead of the Huggins eq. (3), is reported in Figure 2. In this case, we make the substitutions $y = \ln \eta _{rel}/c$

```
POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2
LIST OF MEASURES
TS = 204.30
                 WITHOUT CORRECTIONS
    TIME(SEC) = 563.50
1
                        X(G/DL)= .3212
                                          Y= 5.4738
                                                      Y1= 5.4948
2
    TIME(SEC) = 424.60
                        X(G/DL)= .2141
                                          Y= 5.0357
                                                      Y1= 5.0066
    FIME(SEC) = 361.40
                        X(G/DL)= .1606
                                          Y= 4.7881
3
                                                      ¥1= 4.7626
    TIME(SEC) = 302.40
                        X(G/DL)= .1071
                                          Y= 4.4848
                                                      Y1= 4.5185
STANDARD DEVIATION= .0392
CONFIDENCE INTERVAL = 90
INTERCEPT(=INTRINSIC WISCOSITY)= 4.030 +- .1559
SLOPE= . 4559+01 +- .7223+00
HUGGINS CONSTANT= .2807
                           MAX= .3518
                                        #IN= .2189
```

Fig. 8. OUTPUT of the first program according to cards 31-36 of the INPUT of Table II.

and $a_1 = k'' [\eta]^2$ in eq. (6). The two programs are very similar, they have the same INPUT, and we can change the program only substituting the cards 36, 37, 38, 59, 85, 86, 87, 88, and 89.

Beyond the approximation of eq. (2), another simplification can be done if we know the value of the Huggins constant. This value must be obtained from careful preliminary calibration tests with polymers of identical polydispersity. Equation (6) can now be written as follows:

$$y = a_0 + k' a_0^2 X \tag{10}$$

Indicating with X the value of k'x, we obtain

$$y = a_0 + a_0^2 X (11)$$

where $a_0 = [\eta]$.

	TABLE I
Numerical	Values of Student's t-Coefficient

n - 2	$\alpha/2 = 0.2$	0.1	0.03	0.025	0.01
1	1.376	3.078	6.314	12.71	31.82
2	1.061	1.886	2.920	4.303	6.965
3	0.978	1.638	2.353	3.182	4.541
4	0.941	1.533	2.132	2.776	3.747
5	0.920	1.476	2.015	2.571	3.365
6	0.906	1.440	1.943	2.447	3.143

COMPUTING VISCOSIMETRIC MOLECULAR WEIGHTS 2593

POLYISOBUTYLENE & IN CYCLOHEXANE T=30C VISCOM.1

```
LIST OF MEASUPES
```

```
WITH DENSITY AND KINETIC-ENERGY CORRECTIONS
TS = 323.50
                                   .3125
    TIME(SEC) = 731.10
                        X(G/DL)=
                                           Y= 2.9294
                                                       Y1= 2.9274
1
                                           Y= 3.1760
                                                       Y1= 3.1874
2
    TIME(SEC) = 575.50
                        X(G/DL)=
                                   .2083
    TIME(SEC)= 505.50
                        X(G/DL)=
                                   .1563
                                           Y= 3.3321
                                                       ¥1= 3.3175
٦
a
    TIME(SEC)= 437.20
                        X(G/DL) =
                                   .1042
                                           Y= 3.4423
                                                       Y1= 3.4475
STANDARD DEVIATION= .0137
CONFIDENCE INTERVAL = 95
INTERCEPT(=INTRINSIC VISCOSITY)= 3.708 +- .0803
SLOPE = -. 2497+01 +- . 3824+00
KRAEMER CONSTANT= -- 1816
                             MAX= -.1607
                                            *IN= -.2007
A= .6900+00
                 K= .2769-03
AVERAGE MOL WEIGHT= . 9613+06
                                 MAX= .9917+06
                                                  MIN= .9313+06
```

Fig. 9. OUTPUT of the second program according to cards 3-8 of the INPUT of Table II.

The calculation of a_0 which minimizes the sum of the squares of deviations employs a third-degree equation. In the program of Figure 3, this equation is solved by Cardano formulas using a library subroutine (SEQ 3).

EXAMPLES OF APPLICATION

We report as examples of application the data elaboration of two viscosimetric tests using two different samples of polyisobutylene in cyclohexane at 30°C. The viscometers were chosen in such a way as to evidence the differences of the results obtained according to eqs. (1) and (2). The molecular weights are calculated according to the Krigbaum-Flory relationship¹⁶:

$$[\eta] = 2.76 \times 10^{-4} M^{0.69}$$
 (12)

The experimental data are reported in the INPUT cards. Table II reports the INPUT for both the programs of Figures 1 and 2. The INPUT of the program of Figure 3 is reported in Table III.

1 5 2 1 3 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 4 12.710 4.303 3.182 2.776 2.571 2.447 95 5 0.69 2.76 E-04 0.918 1 7 323.5 731.1 575.5 505.5 437.2 8 9 1 5 10 5 5 10 9 1 10 POLVISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 10 POLVISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 E-04 1 <	Card number			Ca	rd conter	nt		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1	5		·				-
3 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 4 12.710 4.303 3.182 2.776 2.571 2.447 95 5 0.69 2.76 E-04 0.918 6 4 12040 0.764 0.918 7 323.5 731.1 575.5 505.5 437.2 8 0.03125 10 5 5 10 9 1 10 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 10 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 E-04 1 1 1 14 323.5 731.1 575.5 505.5 437.2 13 4 -1 1 1 14 323.5 731.1 575.5 505.7 10 16 1 1 1 1 10	2	1						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3	POLYIS	DBUTYL	ENE A IN	CYCLOH	IEXANE T=	30C VISC	OM.1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4	12.710	4.303	3.182	2.776	2.571	2.447	95
6 4 12040 0.764 0.918 7 323.5 731.1 575.5 505.5 437.2 8 0.03125 10 5 5 10 9 1 10 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 $E-04$ 1 1 1 13 4 -1 1 1 1 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 5 10 16 1 1 1 1 1 17 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 0.69 2.76 E-04 0 20 4 4050 0.764 0.918 90 0.69 2.76 E-04 20 2.358 2.132 2.015 1.943 90 21 204.3 563.5 424.6 361.4 302.4 00 26 0	5	0.0	69	2.7	6 E-04			
7 323.5 731.1 575.5 505.5 437.2 8 0.03125 10 5 5 10 9 1 10 POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1 11 12.710 4.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 $E-04$ 11 1 11 11 11 11 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 510 10 16 1 1 1 1 1 17 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 0.69 2.76 $E-04$ 20 4 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 10 23 1 -1 1 1 1 1 24 POLYISOBUTYLENE B IN CYCLOHEXANE T= $30C$ VISC	6	4	12040)	0.764		0.918	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	7	323.5	731.1	575.5	505.5	437.2		
9110POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.11112.7104.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 $E-04$ 134 -1 1114 323.5 731.1 575.5 505.5 437.2 15 0.03125 1055101611117POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.218 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 $E-04$ 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10551023111124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 $E-04$ 11274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 105510301 -1 11344 -1 1135 204.3 563.5 424.6 361.4 302.4 301 -1 1134 -1 1<	8	0.03125	10	5	5	10		
10POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.11112.7104.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 $E-04$ 13 4 -1 1 1 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 5 10 16 1 1 1 1 17POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.218 6.314 2.920 2.358 2.132 2.015 19 0.69 2.76 $E-04$ 20 4 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 23 1 24 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 26 0.69 2.76 $E-04$ 27 4 -1 1 1 28 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 30 1 -1 1 1 34 4 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 30 1 -1 1 1	9	1						
11 12.710 4.303 3.182 2.776 2.571 2.447 95 12 0.69 2.76 $E.04$ 13 4 -1 1 1 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 5 10 16 1 1 1 1 17 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 $E.04$ 0.918 21 204.3 563.5 424.6 361.4 302.4 20 4 4050 0.764 0.918 21 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 23 1 1 1 1 1 24 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 6.314 2.920	10	POLYISC	BUTYL	ENE A IN	CYCLOF	IEXANE T=	30C VISC	OM .1
12 0.69 2.76 E-04 13 4 -1 1 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 5 10 16 1 1 1 1 17 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 18 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 E-04 0 0.918 11 204.3 563.5 424.6 361.4 302.4 20 4 4050 0.764 0.918 12 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 23 1 1 1 1 10 24 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 25 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.03212 10 5 5 10 1 1 1	11	12.710	4.303	3.182	2.776	2.571	2.447	95
13 4 -1 1 1 14 323.5 731.1 575.5 505.5 437.2 15 0.03125 10 5 5 10 16 1 1 1 1 17 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 18 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 E-04 0 918 21 204.3 563.5 424.6 361.4 302.4 0.918 21 21 204.3 563.5 424.6 361.4 302.4 0.918 21 23 1 1 24 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 25 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 E-04 1 1 1 28 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 1 30 1 1 1	12	0.	6 9	2.7	6 E-04			
14323.5731.1575.5505.5437.2150.0312510551016117POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2186.3142.9202.3582.1322.0151.94390190.692.76E-040.91820440500.7640.91821204.3563.5424.6361.4302.4220.0321210551023111124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2256.3142.9202.3582.1322.0151.94390260.692.76E-0411128204.3563.5424.6361.4302.400290.032121055101301-111131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2326.3142.9202.3582.1322.0151.94390331-11111344-111135204.3563.5424.6361.4302.4360.03212105510	13	4]	1	1		1	
15 0.03125 10551016117POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.218 6.314 2.920 2.358 2.132 2.015 1.943 9019 0.69 2.76 $E-04$ 0.918 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 1055 10 2311 24 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 26 0.69 2.76 $E-04$ 274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 1055 10 3011 31 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 331 -1 1 1 344 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 36 0.03212 10 5 5 10	14	323.5	731.1	575.5	505.5	437.2		
16117POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM 218 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 $E-04$ 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 23124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM 225 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 $E-04$ 274 -1 1 1 28 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 301 1 1 1 31POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM 232 6.314 2.920 2.358 2.132 2.015 1.943 90 331 -1 1 1 34 4 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 0.03212 10 5 5 10	15	0.03125	10	5	5	10		
17POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.218 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 $E-04$ 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 231124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 26 0.69 2.76 $E-04$ 274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 3011 31 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 331 -1 1 1 34 4 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 0.03212 10 5 5 10	16	1						
18 6.314 2.920 2.358 2.132 2.015 1.943 90 19 0.69 2.76 $E-04$ 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 5 5 10 23124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 $E-04$ 274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 3011 31 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90 331 -1 1 1 344 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 36 0.03212 10 5 5 10	17	POLYISC	BUTYL	ENE BIN	CYCLOH	IEXANE T=	30C VISC	OM.2
19 0.69 2.76 $E-04$ 204 4050 0.764 0.918 21 204.3 563.5 424.6 361.4 302.4 22 0.03212 10 55 10 231124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 $E-04$ 274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 55 10 30111131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90 331 -1 11344 -1 1135 204.3 563.5 424.6 361.4 302.4 60.03212 10 55 10	18	6.314	2.920	2.358	2.132	2.015	1.943	90
2044050 0.764 0.918 21204.3563.5424.6361.4302.422 0.03212 10551023124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 9026 0.69 2.76 $E-04$ 274 -1 1128204.3563.5424.6361.4302.429 0.03212 105510301131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90331 -1 11344 -1 11135204.3563.5424.6361.4302.46 0.03212 105510	19	0.	69	2.7	3 E-04			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	4	4050)	0.764		0.918	
22 0.03212 10551023124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 9026 0.69 2.76 $E-04$ 274 -1 1128204.3 563.5 424.6 361.4 302.4 29 0.03212 105510301131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90331 -1 11344 -1 1135204.3 563.5 424.6 361.4 302.4 0.03212 105510	21	204.3	563.5	424.6	361.4	302.4		
23124POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 29 2.358 2132 2.015 26 0.69 27427428 204.3 29 563.5 429 2.76 29 0.03212 10551030131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1 -1 3444 -1 35 204.3 563.5 424.6 36 0.03212 105510	22	0.03212	10	5	5	10		
24POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.225 6.314 2.920 2.358 2.132 2.015 1.943 90 26 0.69 2.76 $E-04$ 274 -1 1128 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 55 10 30131POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90 331 -1 11344 -1 1135 204.3 563.5 424.6 361.4 302.4 0.03212 10 55 10	23	1						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24	POLYISO	BUTYL	ENE B IN	CYCLOR	IEXANE T=	30C VISC	OM.2
26 0.69 2.76 $E-04$ 27 4 -1 1 28 204.3 563.5 424.6 361.4 302.4 29 0.03212 10 5 5 10 30 1 90LYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 32 6.314 2.920 2.358 2.132 2.015 1.943 90 33 1 -1 1 1 34 4 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 36 0.03212 10 5 5 10	25	6.314	2.920	2.358	2.132	2.015	1.943	90
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	0.	69	2.7	6 E-04			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27	4]	L	1		1	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	28	204.3	563.5	424.6	361.4	302.4		
301 31 POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2 32 6.314 2.920 2.358 2.132 2.015 1.943 90 33 1 -1 34 4 -1 1 1 35 204.3 563.5 424.6 361.4 302.4 36 0.03212 10 5 5 10	29	0.03212	10	5	5	10		
31POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.232 6.314 2.920 2.358 2.132 2.015 1.943 90 331 -1 344 -1 1135204.3563.5424.6361.4302.436 0.03212 105510	30	1						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	31	POLYISC	BUTYL	ENE B IN	CYCLOH	IEXANE T=	30C VISC	OM.2
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	32	6.314	2.920	2.358	2.132	2.015	1.943	90
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	33	1		-1				
35 204.3 563.5 424.6 361.4 302.4 36 0.03212 10 5 5 10	34	4	1	L	1		1	
36 0.03212 10 5 5 10	35	204.3	563.5	424.6	361.4	302.4		
	36	0.03212	10	5	5	10		
37 1	37	1						

 TABLE II

 Example of INPUT for the First and Second Programs

 TABLE III

 Example of INPUT for the Third Program

Card number	Card content						
1	2						
2	POLYISOBUTYLENE A IN CYCLOHEXANE T=30C VISCOM.1						
3	4 0.2 0.69 2.76 E-04						
4	323.5 731.1 575.5 505.5 437.2						
5	0.03125 10 5 5 10						
6	1						
7	POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2						
8	4 0.2 0.69 2.76 E-04						
9	204.3 563.5 424.6 361.4 302.4						
10	0.03212 10 5 5 10						
11	1						

COMPUTING VISCOSIMETRIC MOLECULAR WEIGHTS 2595

POLYISOBUTYLENF & IN CYCLOHEXANE T=30C VISCOM.1

LIST OF MEASURES

TS = 323.50 WITHOUT CORRECTIONS

1 TIME(SEC)= 731.10 X(G/DL)= .3125 Y= 2.6091 Y1= 2.6158 2 TIME(SEC)= 575.50 X(G/DL)= .2083 Y= 2.7650 Y1= 2.7622 3 TIME(SEC)= 505.50 X(G/DL)= .1563 Y= 2.8566 Y1= 2.8355 4 TIME(SEC)= 437.20 X(G/DL)= .1042 Y= 2.8914 Y1= 2.9087

STANDARD DEVIATION= .0200

CONFIDENCE INTERVAL = 95

INTERCEPT(=INTRINSIC VISCOSITY)= 3.055 +- .1171

SLOPE = -. 1496+01 +- .5578+00

KRAEMER CONSTANT= -. 1506 MAX= -.0983 MIN= -.1951

A= .6900+00 K= .2760-03

AVERAGE MOL WEIGHT = . 7262+06 MAX = . 7669+06 MIN = . 6862+06 Fig. 10. OUTPUT of the second program according to cards 10-15 of the INPUT of Table II.

POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2

LIST OF MEASURES

TS = 204.30 WITH DENSITY AND KINETIC-ENERGY CORRECTIONS 1 TIME(SEC)= 563.50 X(G/DL)= .3212 Y= 3.4386 Y1= 3.4291 2 TIME(SEC)= 424.60 X(G/DL)= .2141 Y= 3.7890 Y1= 3.8068 3 TIME(SEC)= 361.40 X(G/DL)= .1606 Y= 3.9932 Y1= 3.9956 4 TIME(SEC)= 302.40 X(G/DL)= .1071 Y= 4.1951 Y1= 4.1844

STANDARD DEVIATION= .0162

CONFIDENCE INTERVAL = 90

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Fig. 11 (continued)
INTERCEPT(=INTRINSIC VISCOSITY)= 4.562 +- .0645
SLOPE= -. 3527+01 +- .2988+00
KRAEMER CONSTANT= -.1695 MAX= -.1596
                                          MIN= -.1787
A= .6900+00
                K= .2760-03
AVERAGE HOL WEIGHT= .1298+07 MAX= .1325+07
                                                 MIN= .1272+07
Fig. 11. OUTPUT of the second program according to cards 17–22 of the INPUT of Table II.
    POLVISOBUTHLENE B IN CYCLOHEXANE T=30C VISCOM.2
LIST OF MEASURES
 TS = 204.30
                 WITHOUT CORRECTIONS
   TIME(SEC) = 563.50 X(G/DL) = .3212
 1
                                          Y= 3.1587
                                                      ¥1= 3.1613
   TIME(SEC)= 424.60 X(G/DL)= .2141
 2
                                          Y= 3.4164
                                                      ¥1= 3.4156
    TIME(SEC)= 361.40
                                                      ¥1= 3.5427
 3
                       X(G/DL)= .1606
                                          Y= 3.5517
    TIME(SEC)= 302.40 X(G/DL)= .1071 Y= 3.6628
                                                      ¥1= 3.6699
 4
STANDARD DEVIATION= .0083
CONFIDENCE INTERVAL = 90
INTERCEPT(=INTRINSIC VISCOSITY)= 3.924 +- .0330
SLOPE = -. 2375+01 +- .1528+00
KRAEMER CONSTANT= -. 1542 MAX= -. 1468 MIN= -. 1614
A⊐ .6900+00
                K= .2760-03
                                MAX= .1056+07 MIN= .1031+07
AVERAGE MOL WEIGHT= .1044+07
Fig. 12. OUTPUT of the second program according to cards 24-29 of the INPUT of Table II.
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2596
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POLYISOBUTYLENE B IN CYCLOHEXANE T=30C VISCOM.2

LIST OF MEASURES

TS = 204.30 WITHOUT CORRECTIONS

TIME(SEC) = 563.50 1 X(G/DL)= .3212 Y= 3.1587 Y1= 3.1613 5 TINE(SEC) = 424.60 X(G/DL)= .2141 Y= 3.4164 ¥1= 3.4156 TIME(SEC)= 361.40 X(G/DL)= .1606 Y= 3.5517 ¥1= 3.5427 3 TIME(SEC) = 302.40 X(G/DL)= .1071 Y= 3.6628 ¥1= 3.6699 .

STANDARD DEVIATION= .0083

CONFIDENCE INTERVAL = 90

INTERCEPT(=INTRINSIC VISCOSITY)= 3.924 +- .0330

SLOPE = -. 2375+01 +- .1528+00

KRAEMER CONSTANT: -.1542 MAX: -.1468 MIN: -.1614 Fig. 13. OUTPUT of the second program according to cards 31-36 of the INPUT of Table II.

POLYISOBUTYLENE & IN CYCLOHEXANE TEBOC VISCON.1

LIST OF MEASURES

75= 323.53 HUBSINS CONSTANT= . 2090 1 TIME(SEC)= 731.10 C(G/DL)= .3125 X(-PK+C)= .3625 Y= 4.0319 Y1= 3.9434, DIF= .0885 TIME(SEC)= 575.50 C(G/DL)= .2083 X(=PK+C)= .0417 Y= 3.7391 Y1= 3.7201 2 DIF= .0190 3 TINE(SEC)= 505,50 C(G/DL)= +1563 X(TPK+C)= +3312 ¥= 3.6906 Y1= 3+60.85 DIF= -.0079 TIME(SEC)= 437.20 C(G/DL)= .1042 X(:PK+C): .0208 Y= 3.3741 Y1= 3.4968 DIF= -.1227 . NUMBER OF REAL ROOTS:3 AZEROI-INTERCEPT-INTRINSIC VISCOSTIVI- 3.273/ A= .6400 K= .276-03 MOL WEIGHT= .8026+06

Fig. 14. OUTPUT of the third program according to cards 2-5 of the INPUT of Table III.

We report the OUTPUT of the first program, which employs the Huggins equation, in Figures 4-8; the OUTPUT of the second program, which employs the Kraemer equation, in Figures 9-13; the OUTPUT of the third program, in Figures 14 and 15.

POLYISOBUTYLENE P IN CYCLOHEXANE T=30C VISCOM-2

LIST OF MEASURES

TS= 204.30 HUGGINS CONSTANT= .2000 1 TIME(SEC)= 563.50 C(6/DL)= .3212 X(-PK+C)= . 0642 Y: 5.4738 Y1= 5.3851 DIF= .0887 2 TIME(SEC)= 424.60 C(G/DL)= DIF= .0344 .2141 X(:PK+C): .1428 ¥ = 5-0357 Y1= 5.0013 TTHE(SFC)= 361.40 DIF= -.0213 3 C(6/DL)= -1606 X(=PK+C)= - 0321 7881 Y1= 4.8094 TINE(SEC)= 302,40 C(G/DL)= .1071 x(:pK+C)= .3214 Y = 8.8888 ¥1= 4+6175 DIF= --1327 NURBER OF REAL ROOTS=3 AZERO(=INTERCEPT=INTRINSIC VISCOSITY) = 4.2337 A= .6700 K= .276-03 HOL WEIGHT: +1165+07

Fig. 15. OUTPUT of the third program according to cards 7-10 of the INPUT of Table III.

In cards 4, 11, 18, 25, and 32 of the INPUT of Figure 2, the Student's *t*-coefficients (already tabulated in Table I) for the desired confidence interval (last number on the right) up to 8 experimental points are reported.

Cards 5, 12, 19, 26, and 33 report the a and K values of eq. (5) for the considered polymer-solvent system: in our case, the data of eq. (12). When the INPUT value for K is negative (especially 33), the program does not calculate the molecular weight.

Cards 6, 13, 20, 27, and 34 report the number of experimental points, the viscometer constant K of eq. (2) (if we write a negative value for K, eq. (3) is employed), therefore the solvent and polymer densities.

Cards 7, 14, 21, 28, and 35 report the solvent and solutions efflux times.

Cards 8, 15, 22, 29, and 36 report first the polymer weight, the initial solvent volume (in cc), and then the solvent volume added at every dilution.

For the same polymer it is possible to place cards of the type 6, 13, etc., after cards of the type 8, 15, etc., obtaining a single list of results.

With regard to the cards reported in the INPUT of Table III, only cards 3 and 8 are different from those of Table II. In these cards, the number of experimental points, the Huggins constant, the a and K values of eq. (12) are reported.

References

1. M. L. Huggins, J. Amer. Chem. Soc., 64, 2716 (1942).

2. E. O. Kraemer, Ind. Eng. Chem., 30, 1200 (1938).

H. Mark, Die Feste Körper, Ed. Akademische Verlag, Lipsia, 1938.

4. R. Houwink J. Prakt. Chem. 175, 15 (1940).

5. P. A. D. De Maine and R. D. Seawright, Digital Computer Programs for Physical Chemistry, MacMillan, New York, 1963, p. 394.

6. I. Klein and D. I. Marshall, Computer Programs for Plastic Engineers, SPE, Reinhold, New York, 1968, p. 349.

7. M. Pegoraro, A. Penati, J. Barês, and A. Ascheri, Chim. Ind. (Milan), 53, 453 (1971).

8. B. T. Hofeiter, J. O. Ernst, and W. L. Williams, J. Appl. Polym. Sci., 17, 1449 (1973).

9. Ch'ien Jên-Yüan, Determination of Molecular Weights of High Polymers, IPST, Jerusalem, 1963, p. 47.

10. J. B. Kinsinger in *Encyclopedia of Polymer Science and Technology*, Vol. 14, H. F. Mark, N. G. Gaylord, and N. Bikahes, Eds., Interscience, New York, 1971, p. 717.

11. P. J. Flory, Principles of Polymer Chemistry, 8th ed., Cornell University Press, Ithaca, N.Y., 1971, p. 309.

12. R. H. Wagner, Anal. Chem. 20, 155 (1948).

13. G. V. Schulz, Z. Elektrochem., 43, 479 (1937).

14. A. H. Bowker and G. J. Lieberman, Engineering Statistics, Prentice Hall, New Jersey, 1959, p. 247.

15. A. H. Bowker and G. J. Lieberman, *ibid.* p. 558.

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